

U-Th-Pb Geochronology

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Corso di formazione ambientale:
Geocronologia: tecniche di datazione e metodi di correlazione

Back to basics

U-Th-Pb age equations

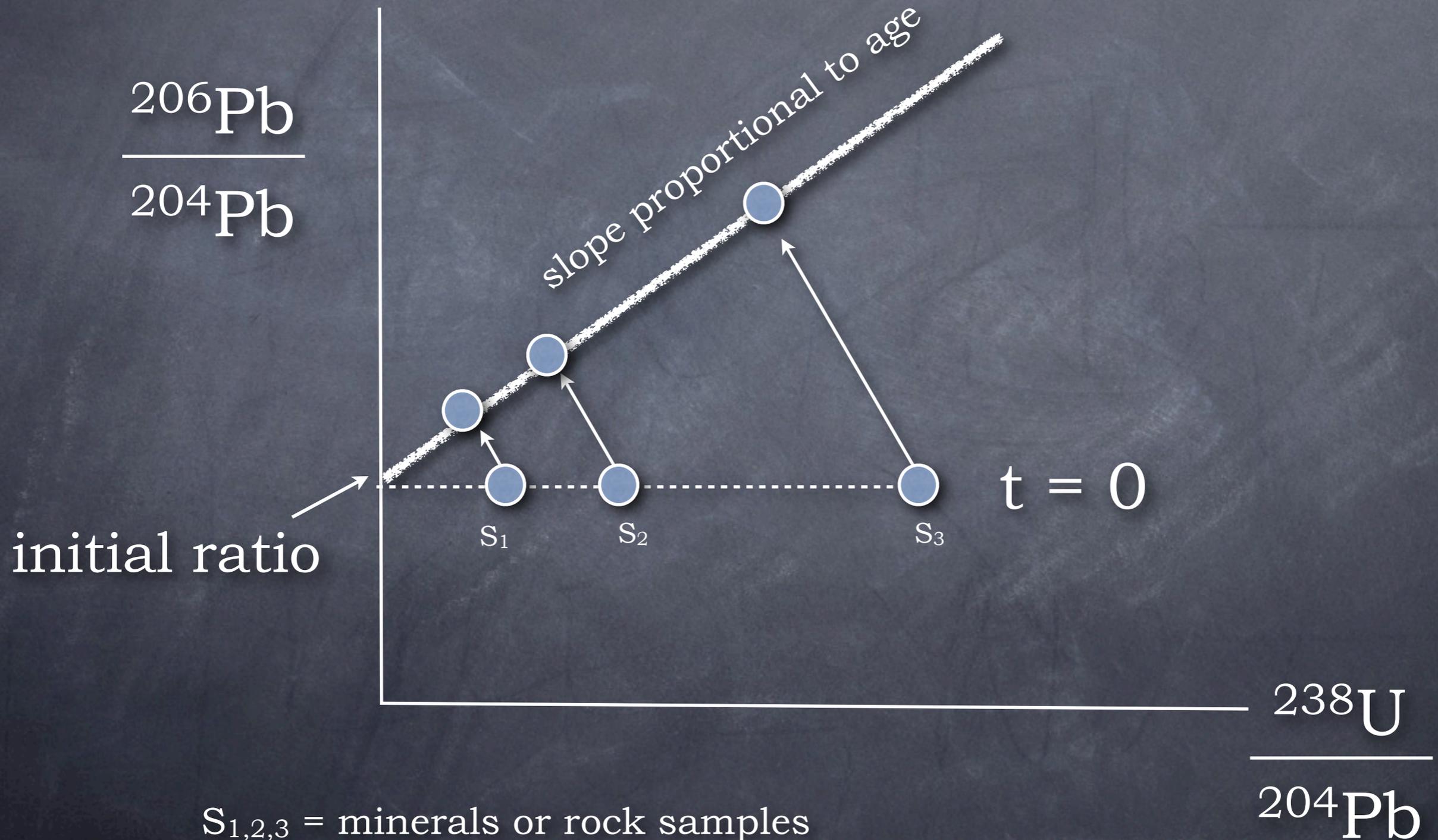
$$\frac{{}^{206}\text{Pb}}{{}^{204}\text{Pb}} = \left(\frac{{}^{206}\text{Pb}}{{}^{204}\text{Pb}} \right)_I + \frac{{}^{238}\text{U}}{{}^{204}\text{Pb}} \left(e^{\lambda t} - 1 \right)$$

$$\frac{{}^{207}\text{Pb}}{{}^{204}\text{Pb}} = \left(\frac{{}^{207}\text{Pb}}{{}^{204}\text{Pb}} \right)_I + \frac{{}^{235}\text{U}}{{}^{204}\text{Pb}} \left(e^{\lambda t} - 1 \right)$$

$$\frac{{}^{208}\text{Pb}}{{}^{204}\text{Pb}} = \left(\frac{{}^{208}\text{Pb}}{{}^{204}\text{Pb}} \right)_I + \frac{{}^{232}\text{Th}}{{}^{204}\text{Pb}} \left(e^{\lambda t} - 1 \right)$$

How to solve age equations?

Isochron age



if initial Pb = 0

$$\frac{{}^{206}\text{Pb}}{{}^{204}\text{Pb}} = \left(\frac{{}^{206}\text{Pb}}{{}^{204}\text{Pb}} \right)_I + \frac{{}^{238}\text{U}}{{}^{204}\text{Pb}} (e^{\lambda t} - 1)$$



$$\frac{{}^{206}\text{Pb}}{{}^{238}\text{U}} = (e^{\lambda t} - 1)$$

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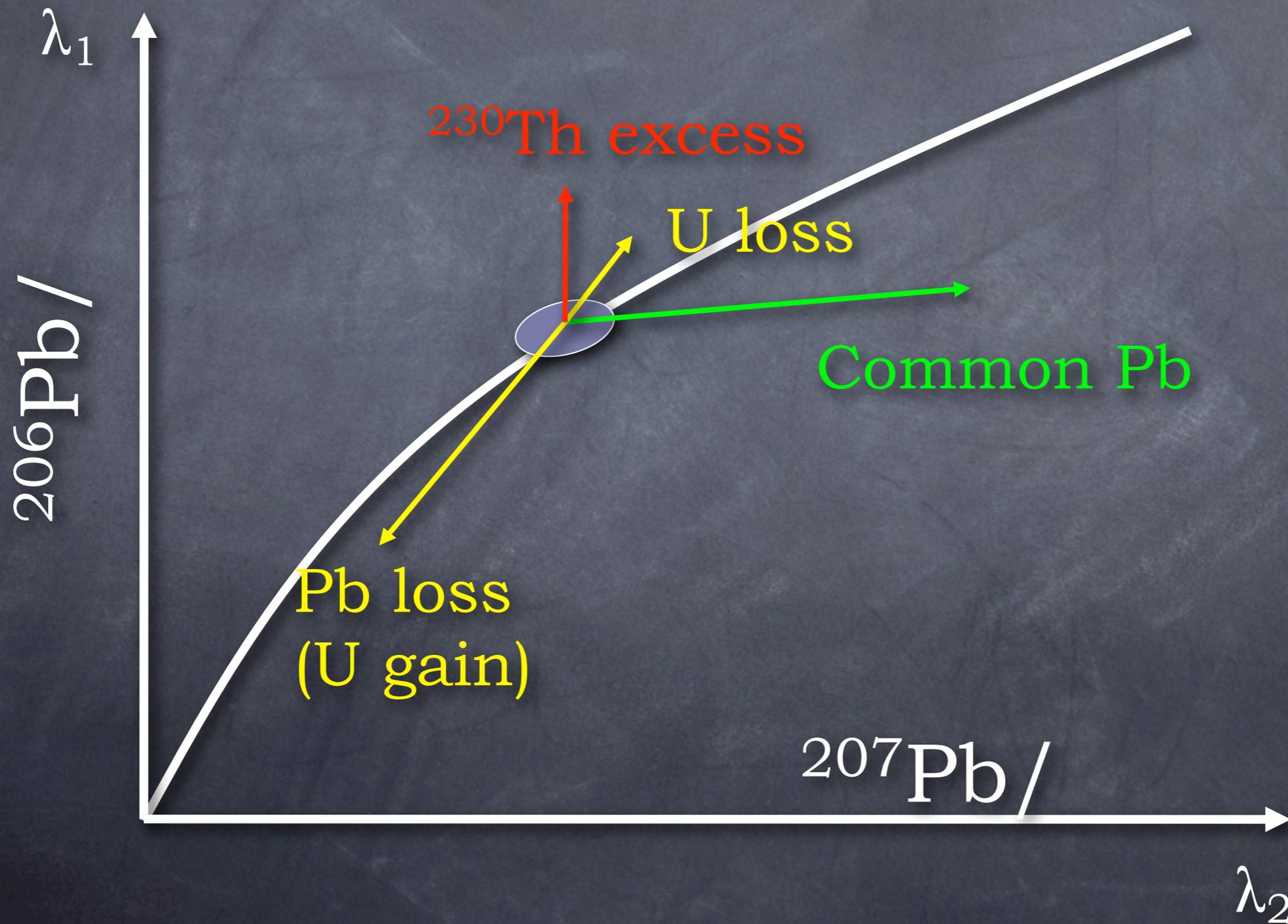
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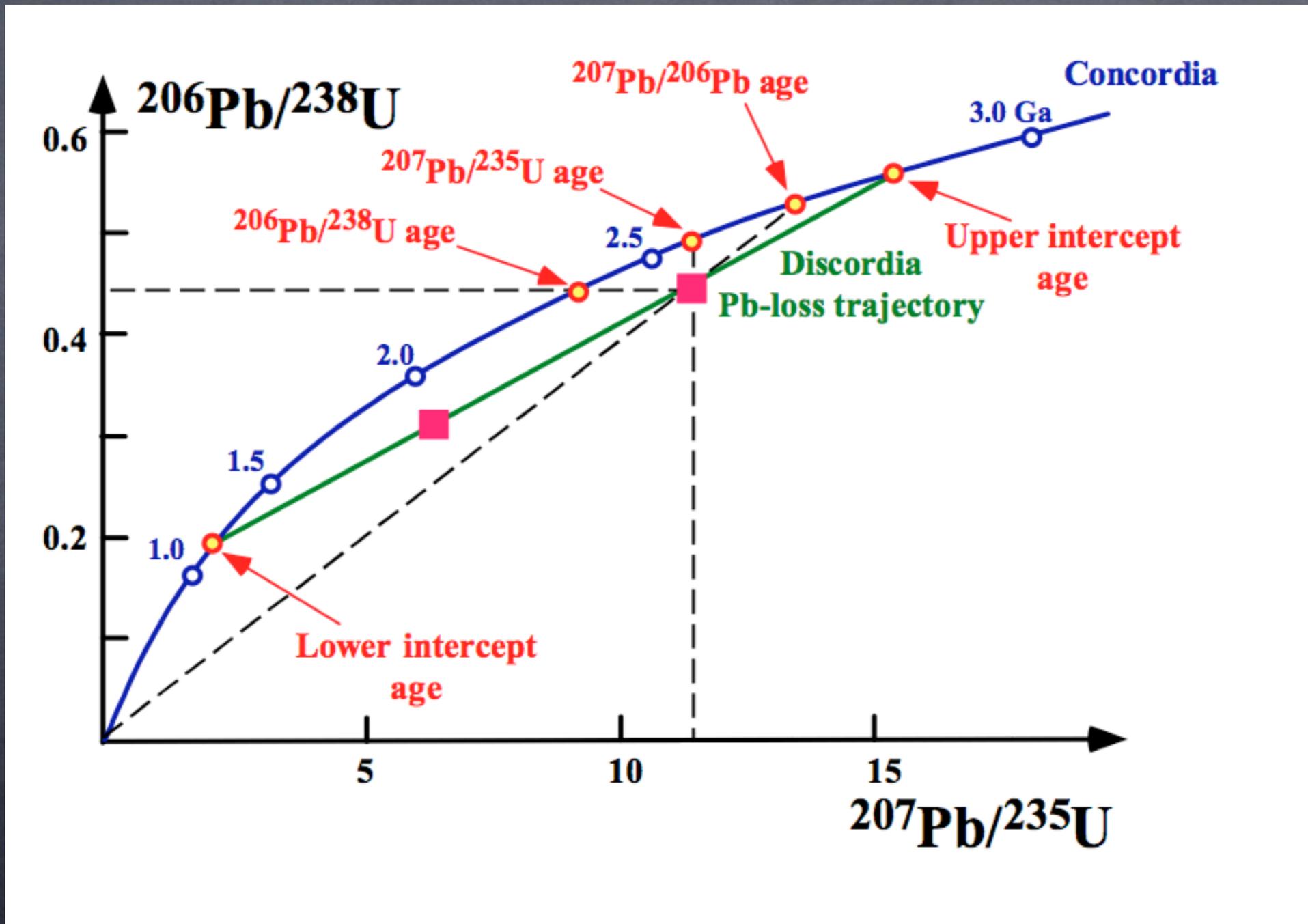


$$\frac{{}^{208}\text{Pb}}{{}^{232}\text{U}} = (e^{\lambda t} - 1)$$

The concordia plot



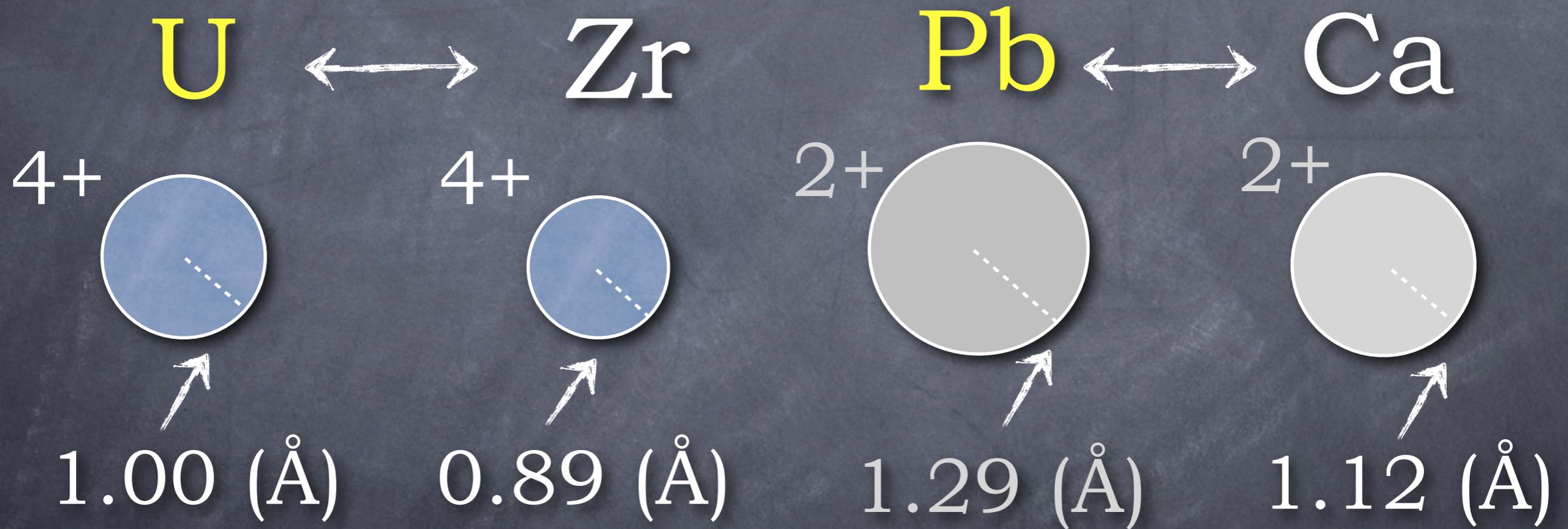
Discordant data



Suitable

high U or Th
low initial Pb

When Pb_0 is low?



in highly selective structures

Mineral	U content (ppm)	Common Pb (% of total Pb)
Zircon	1 to >10 000	<2%
Monazite	282 to >50 000	<2%
Baddeleyite	58 to 3410	<2%
Rutile	<1 to 390	<2 to 95%
Xenotime	5000 to 29 000	<5%
Titanite	4 to 500	5 to 40%
Allanite	130 to 600	5 to 30%

Data from Heaman and Parrish (1991), Parrish and Tarrul (1989) and Noble and Searle (1995)

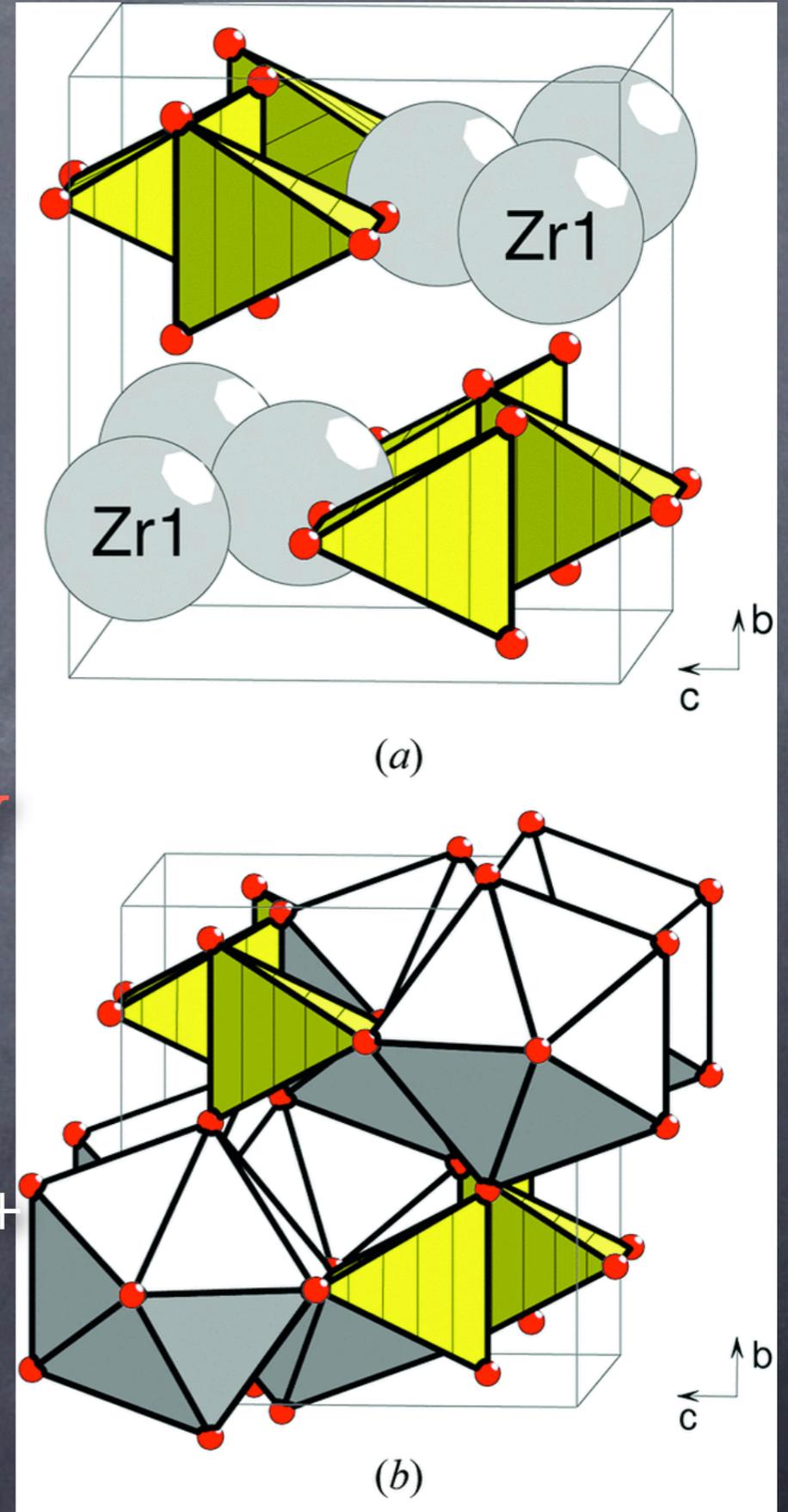


Zircon (tetragonal)



$[8]A[4]TO_4$ stoichiometry

$[8]Zr^{4+}$, (HREE $^{3+}$), U^{4+} , Th^{4+}



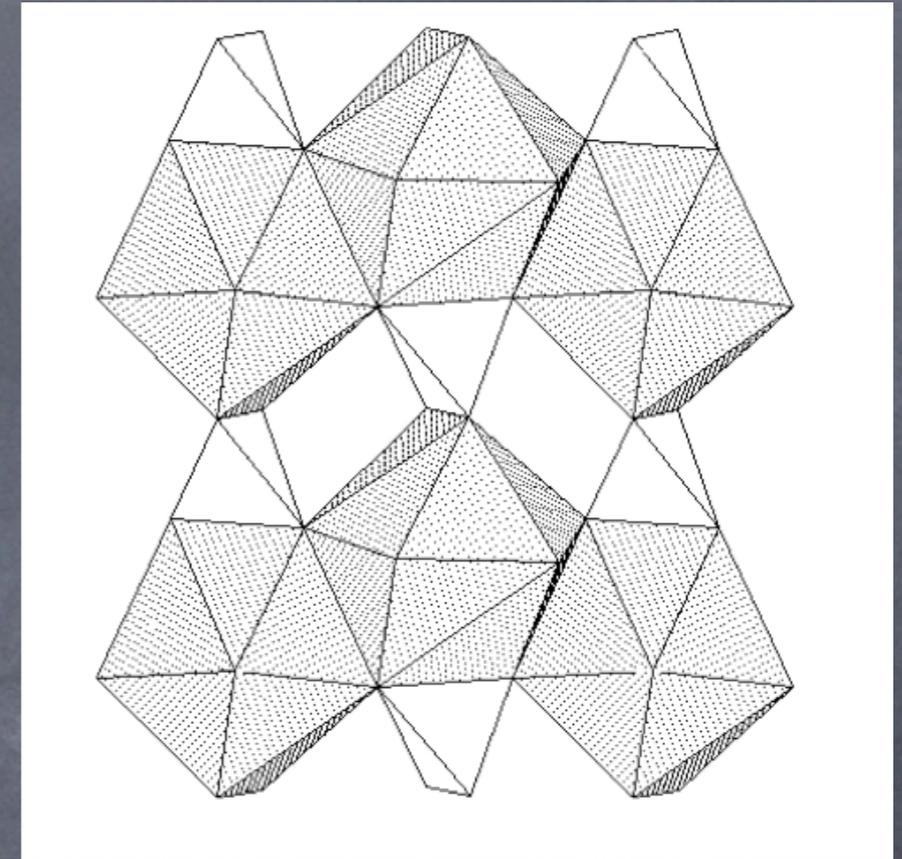
Monazite (monoclinic)

[4]P⁵⁺

[8]A^[4]TO₄ stoichiometry

LREE³⁺, Th⁴⁺

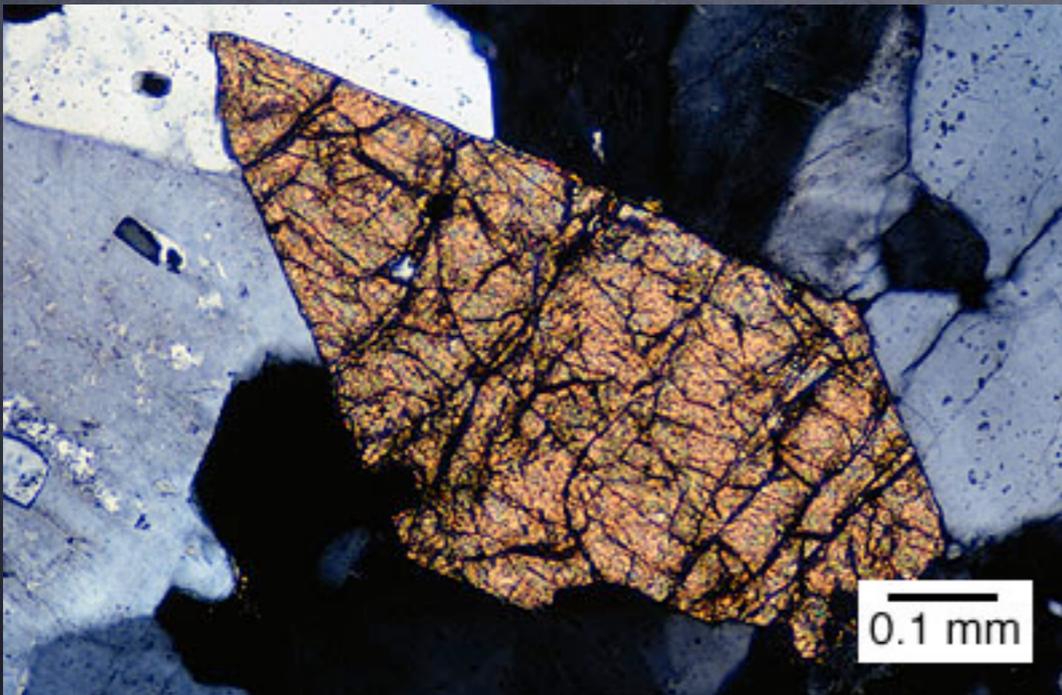
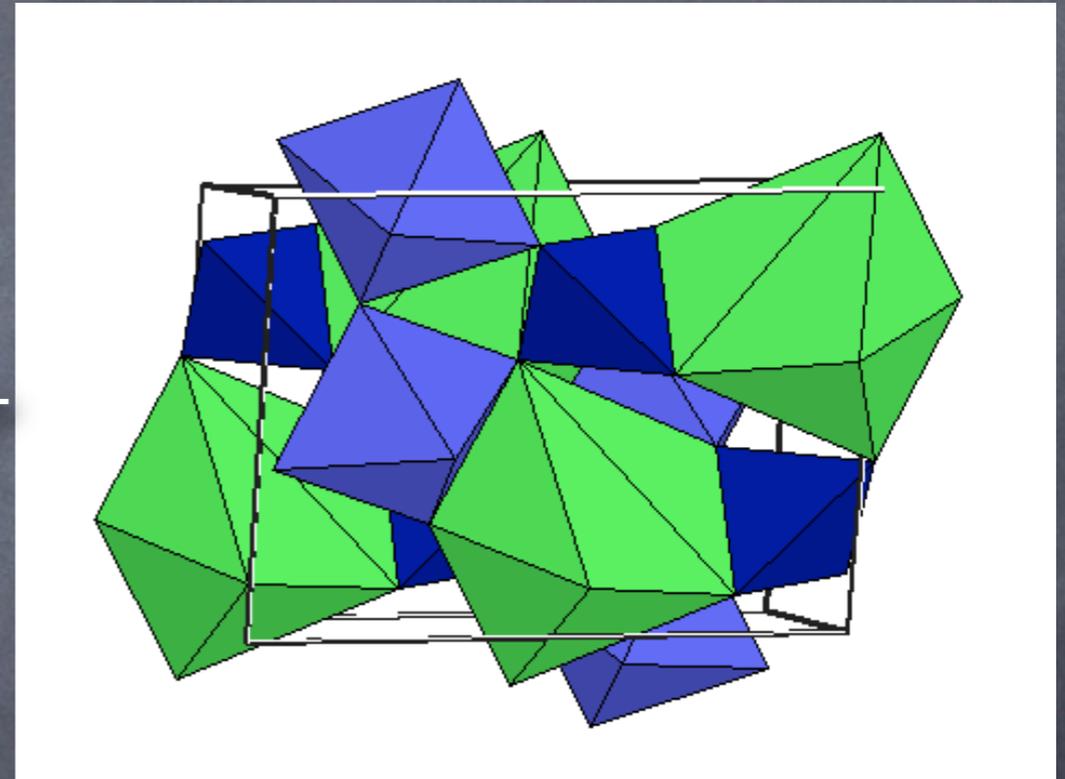
(U⁴⁺, Ca²⁺, Sr²⁺, Pb²⁺, Ba²⁺)



Titanite



LREE³⁺, U⁴⁺, Th⁴⁺, Pb²⁺



Different

different P, T, x stability fields

different closure temperatures

different ages

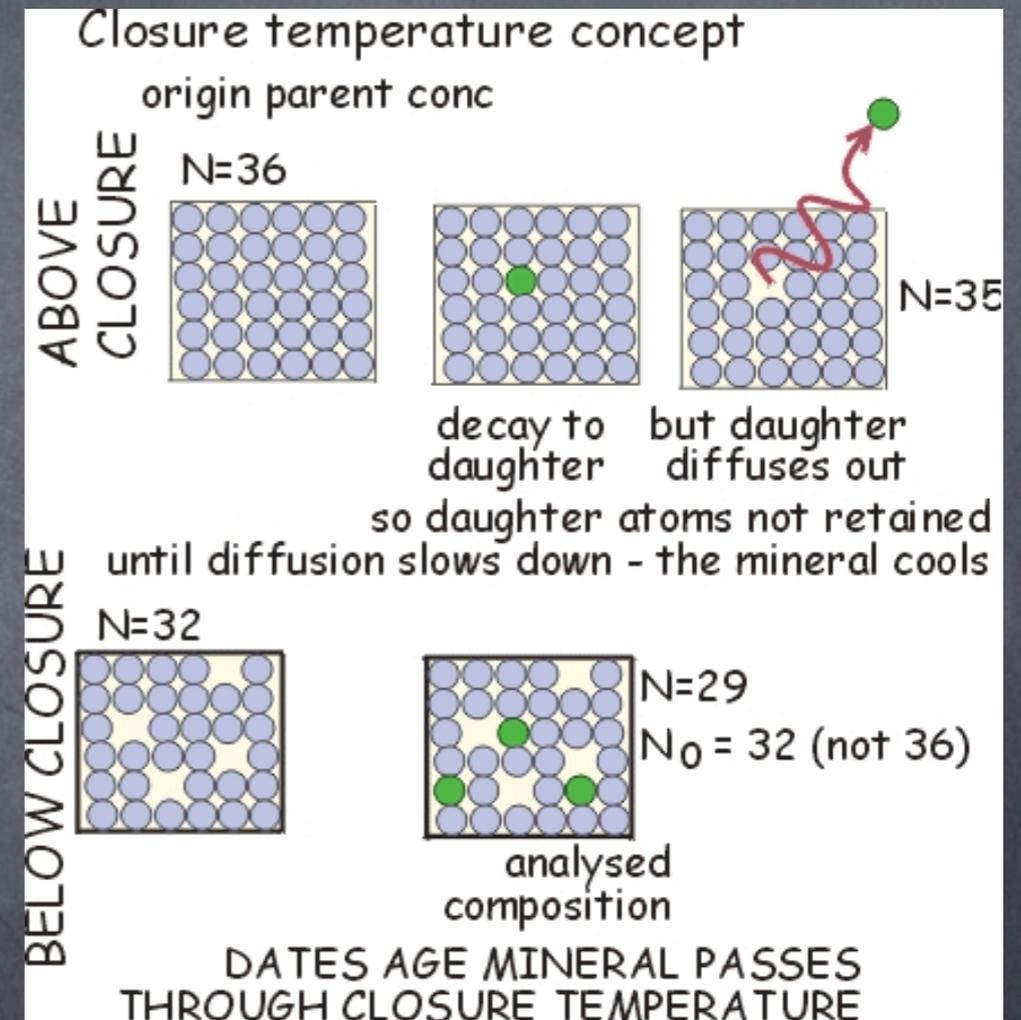
“What is closing
temperature???” “

“Closing temperature is an
individual concept !!!”

closure temperature

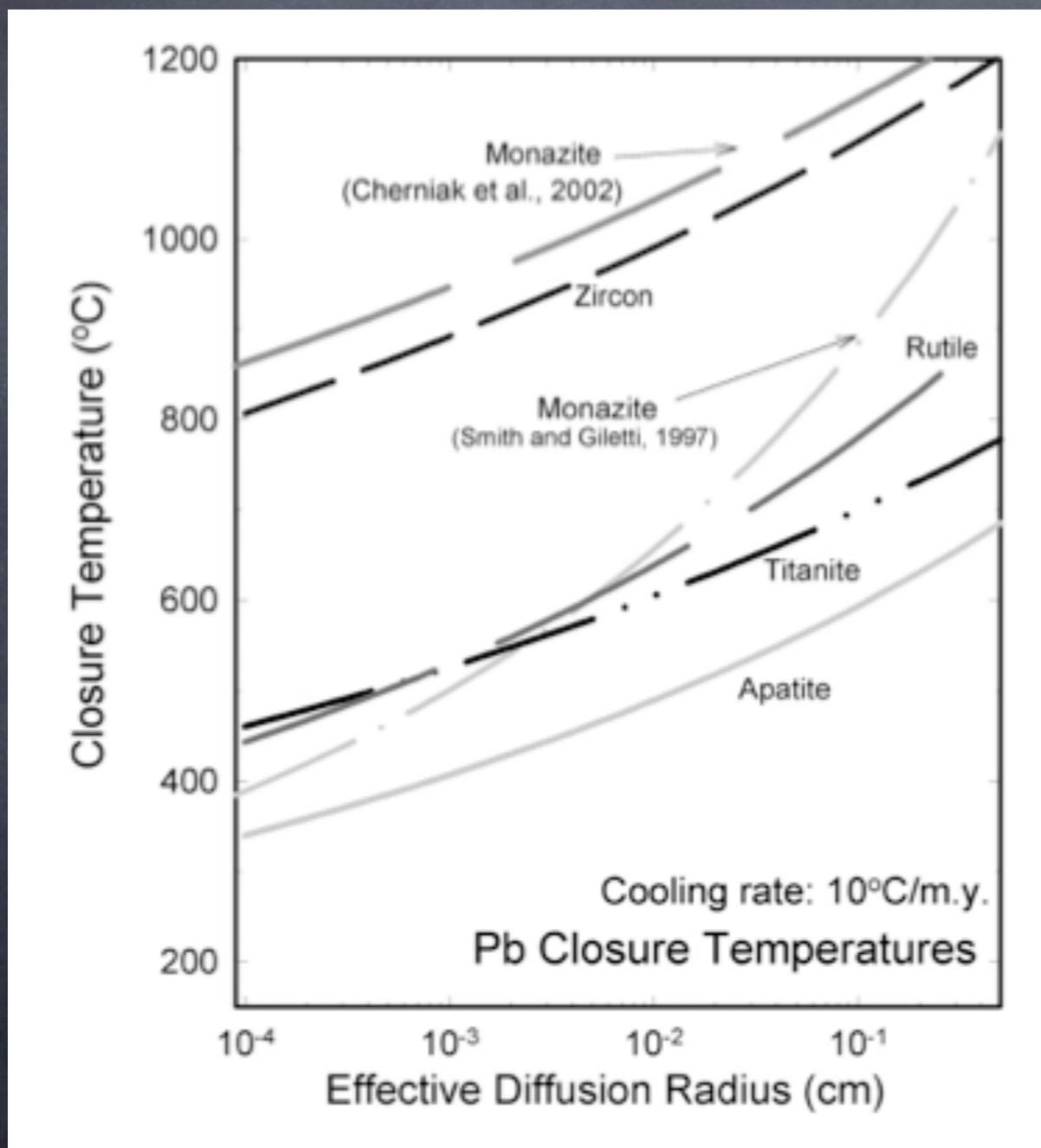
The closure temperature of a geochronological system may be defined as the temperature at the time corresponding to the apparent age...

The temperature recorded by a "frozen" chemical system, in which a solid phase in contact with a large reservoir has cooled slowly from high temperatures, is formally identical with geochronological closing temperature

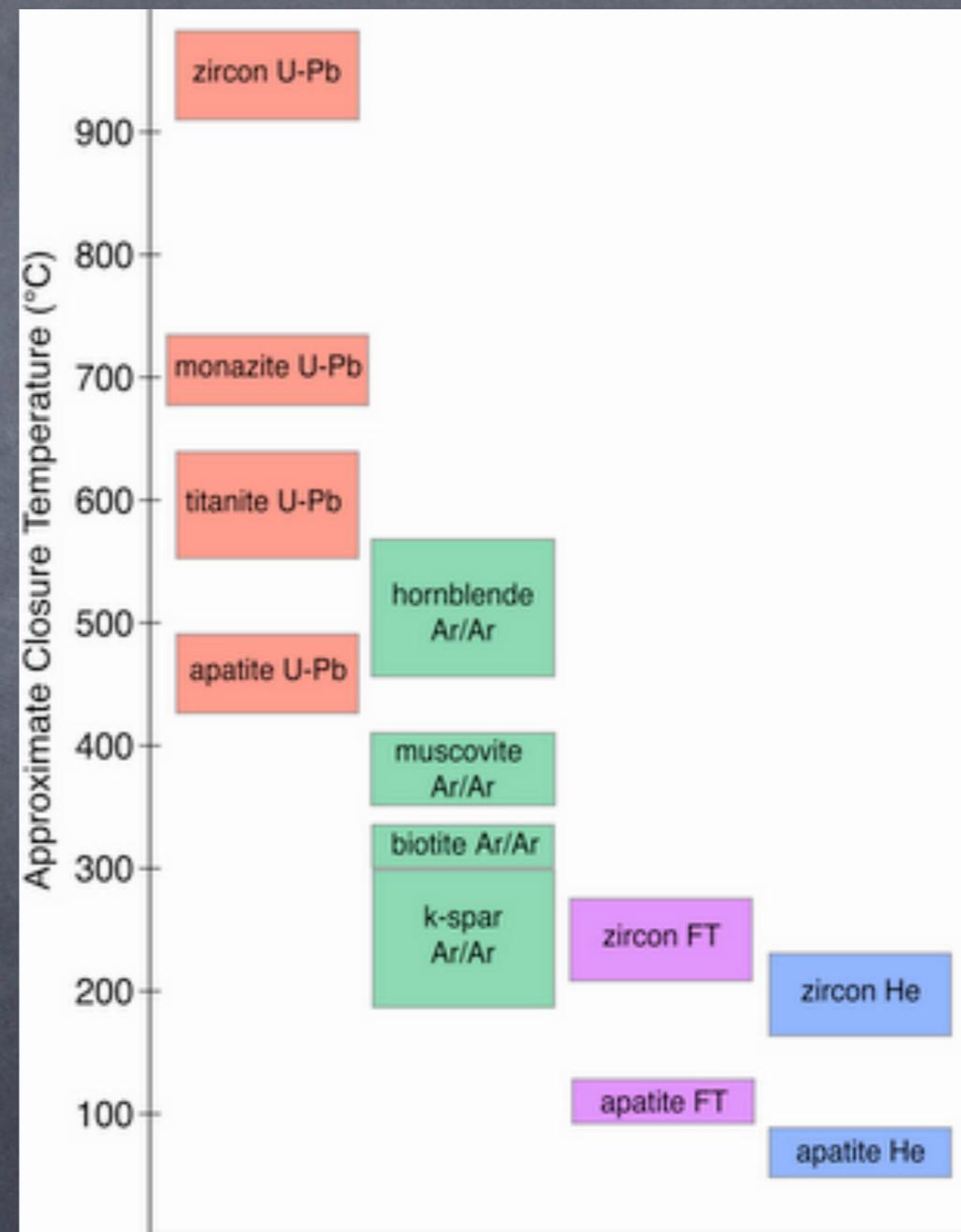


Once a mineral cools enough, then the daughter product can accumulate, and the radiometric "clock" has started.

T_c vs. minerals

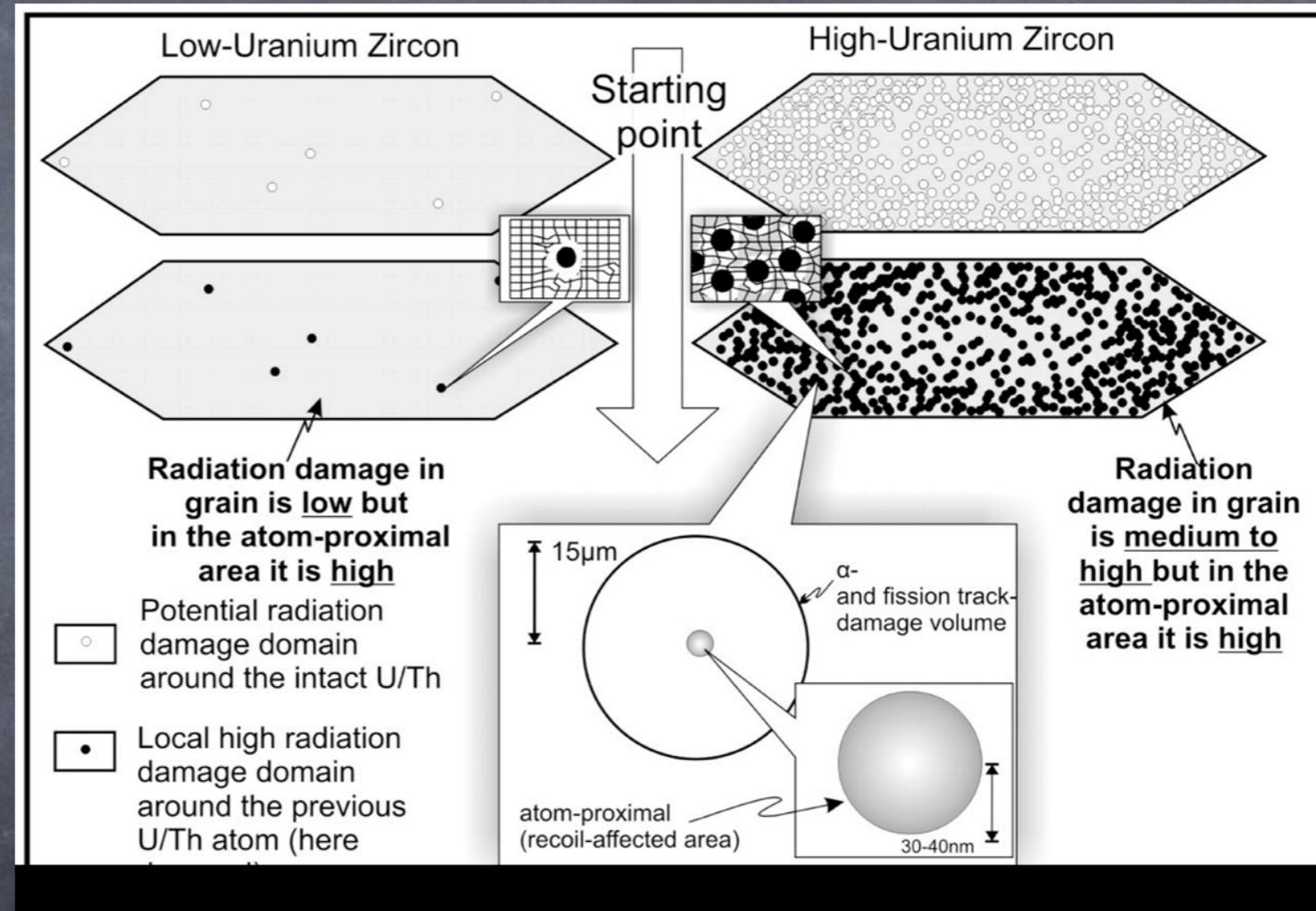
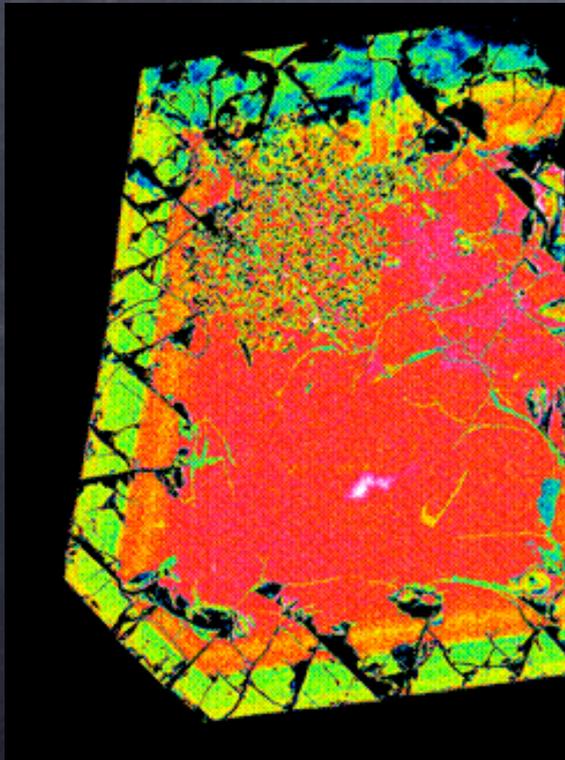
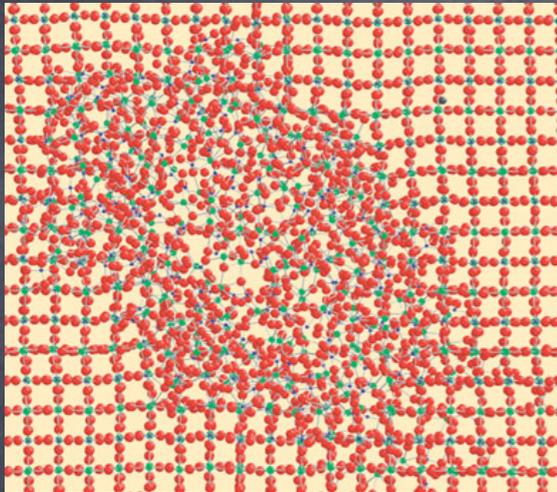


Cherniak & Watson, 2003



after Dodson (1973),
Contrib. Min. Pet., 40, 259-274

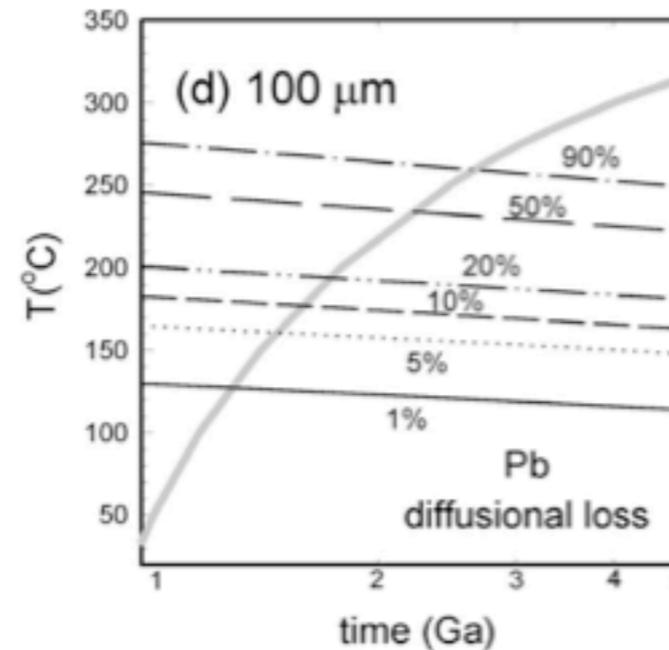
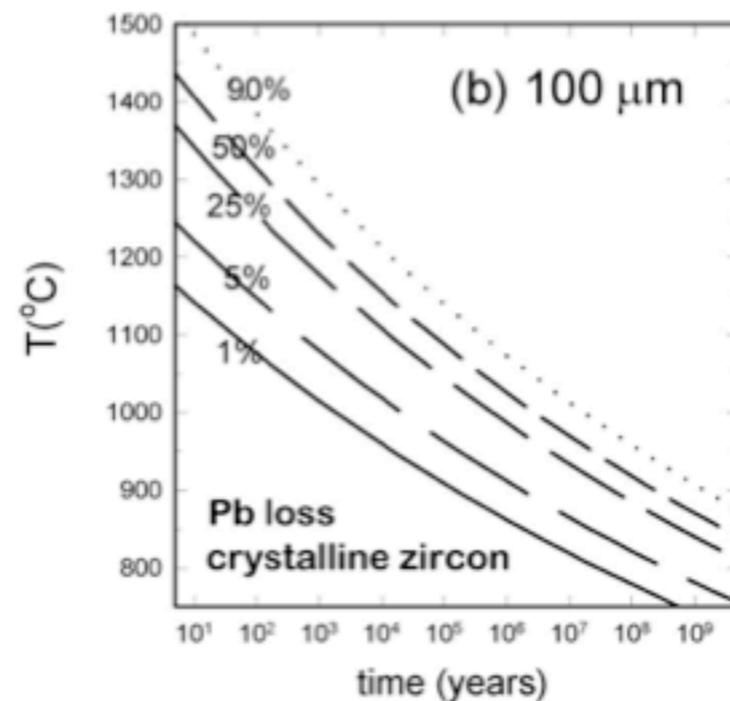
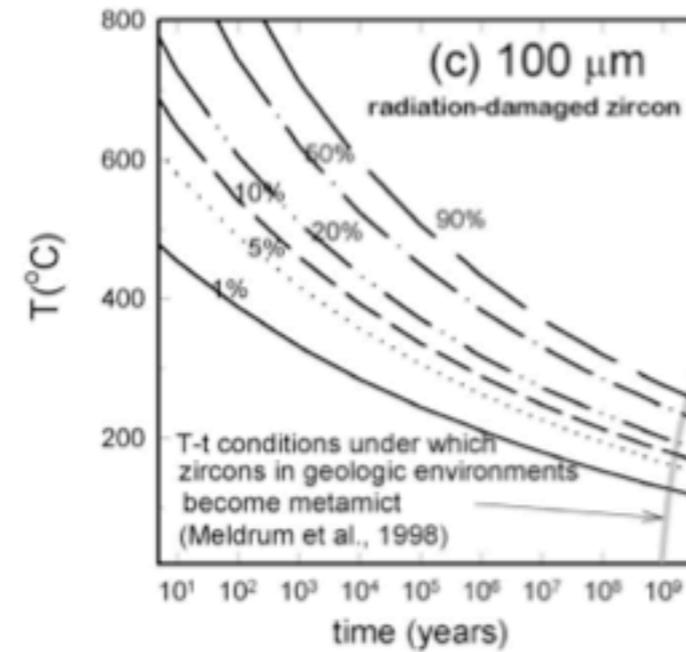
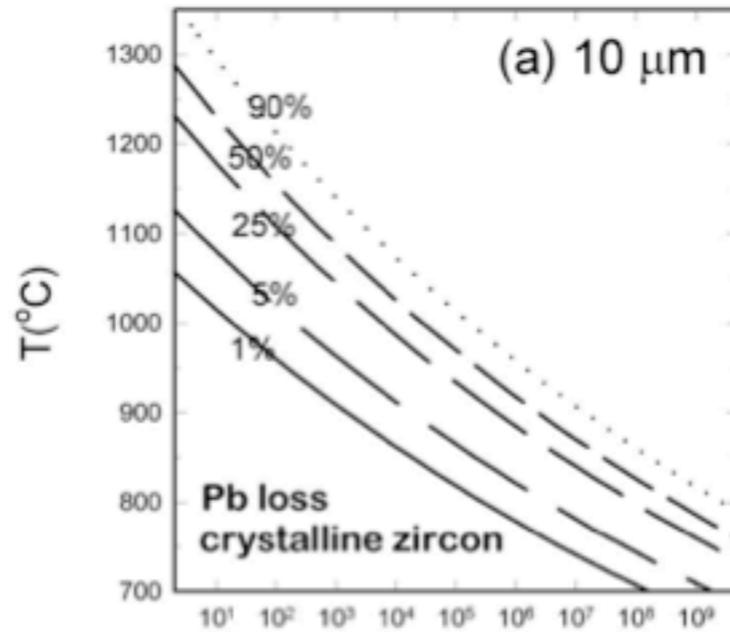
High U = radiation damage



Marsellos and Garver *American Mineralogist*. 2010; 95: 1192-1201

→ Pb loss

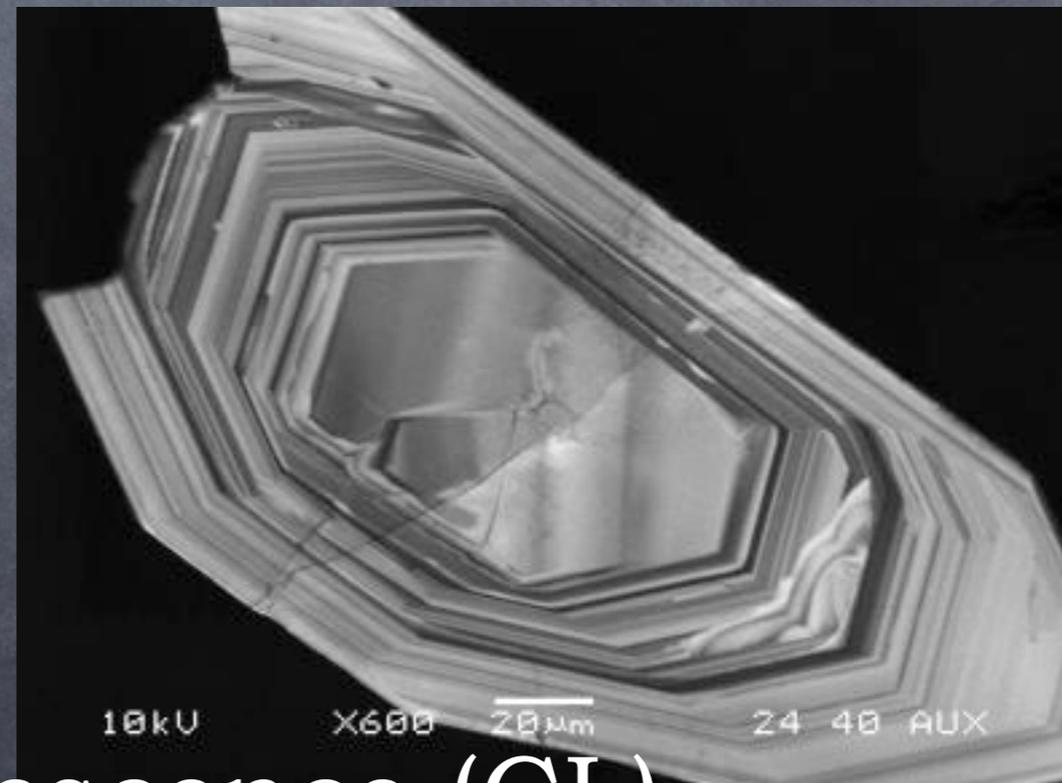
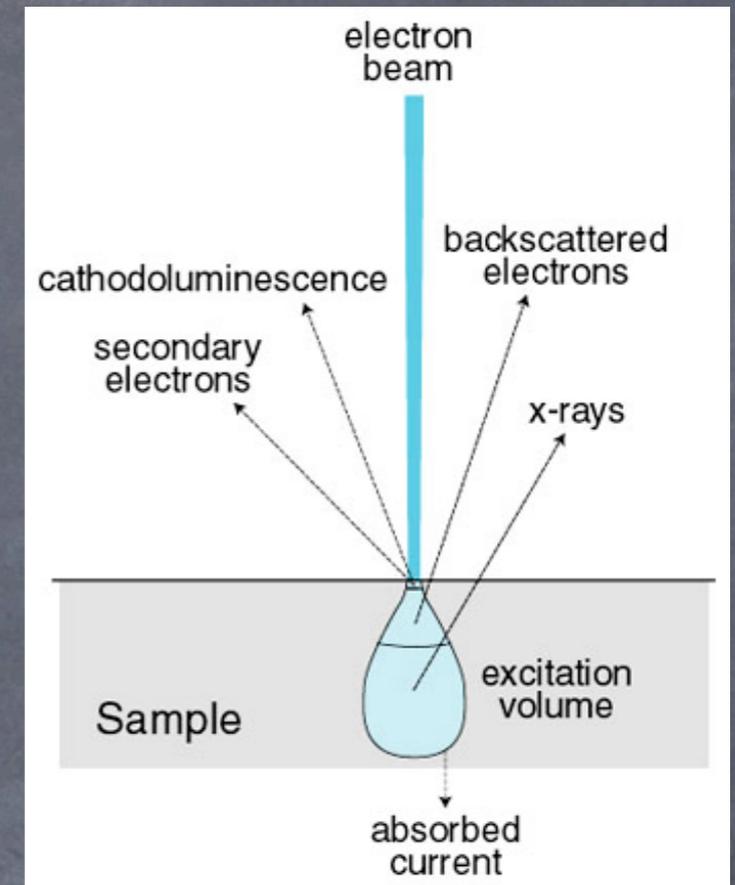
Pb loss in zircon



Zircon structure



Backscattered image

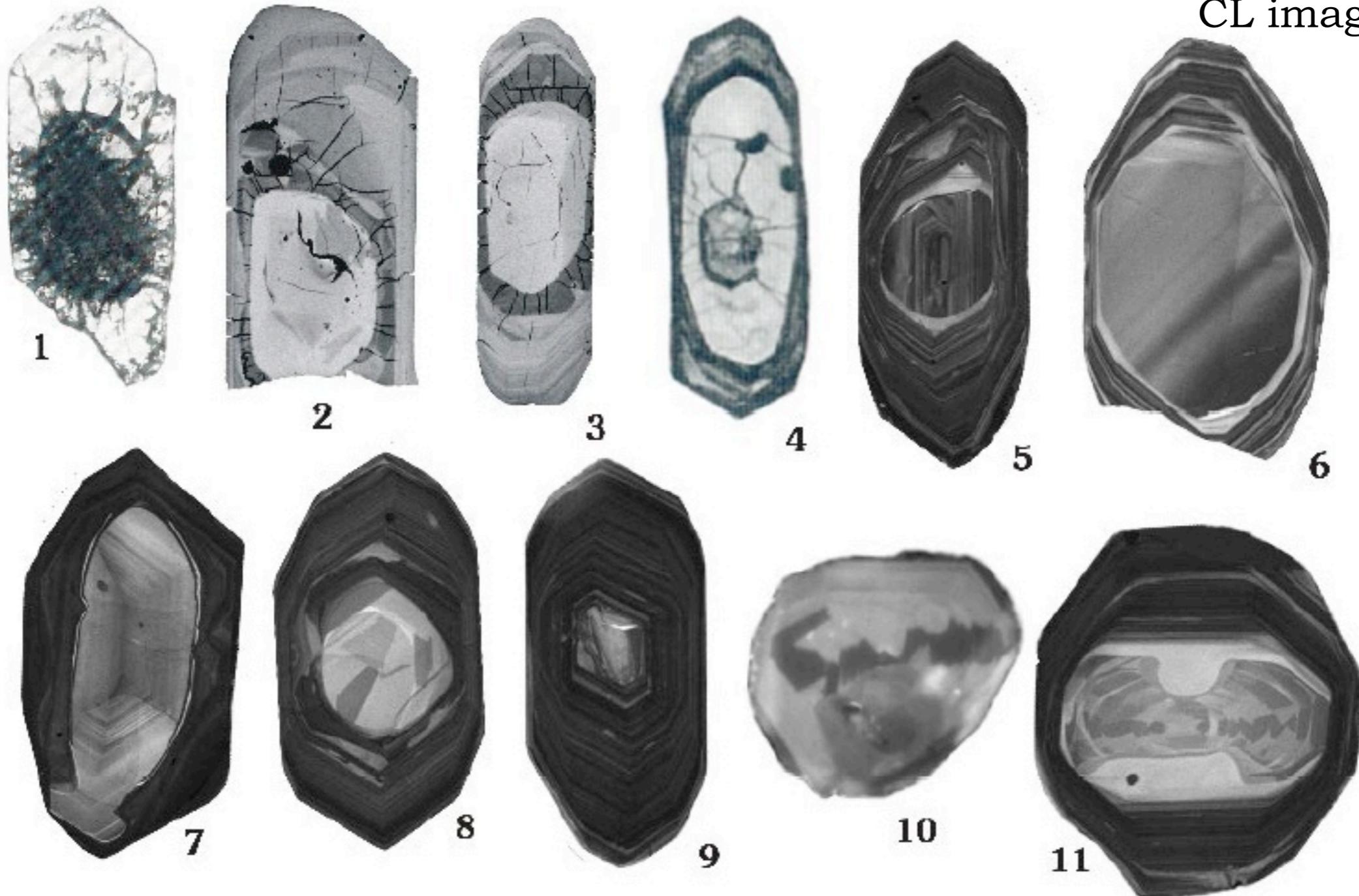


cathodoluminescence (CL)
image

Zircon

Corfu et al., RiMG 2003

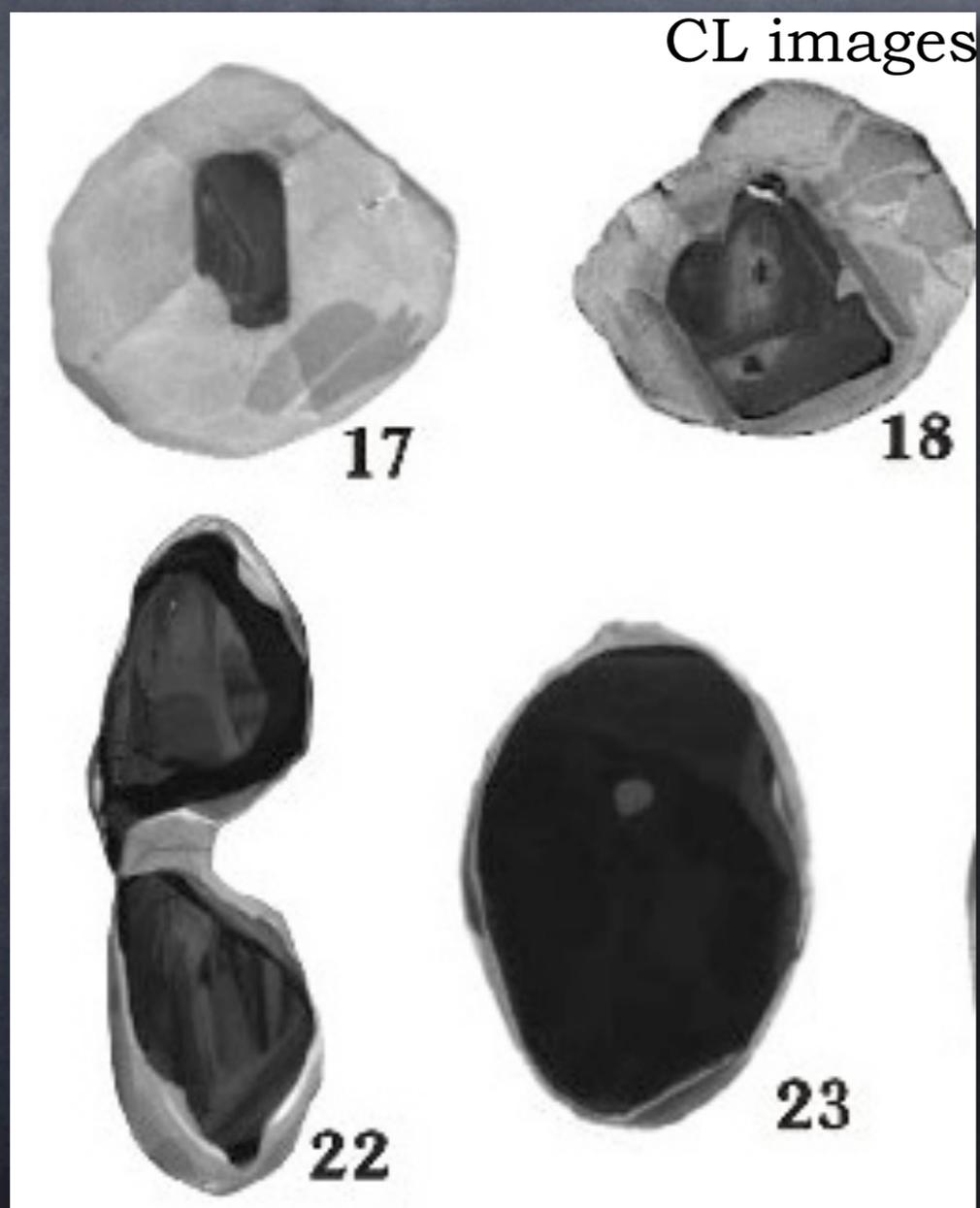
CL images



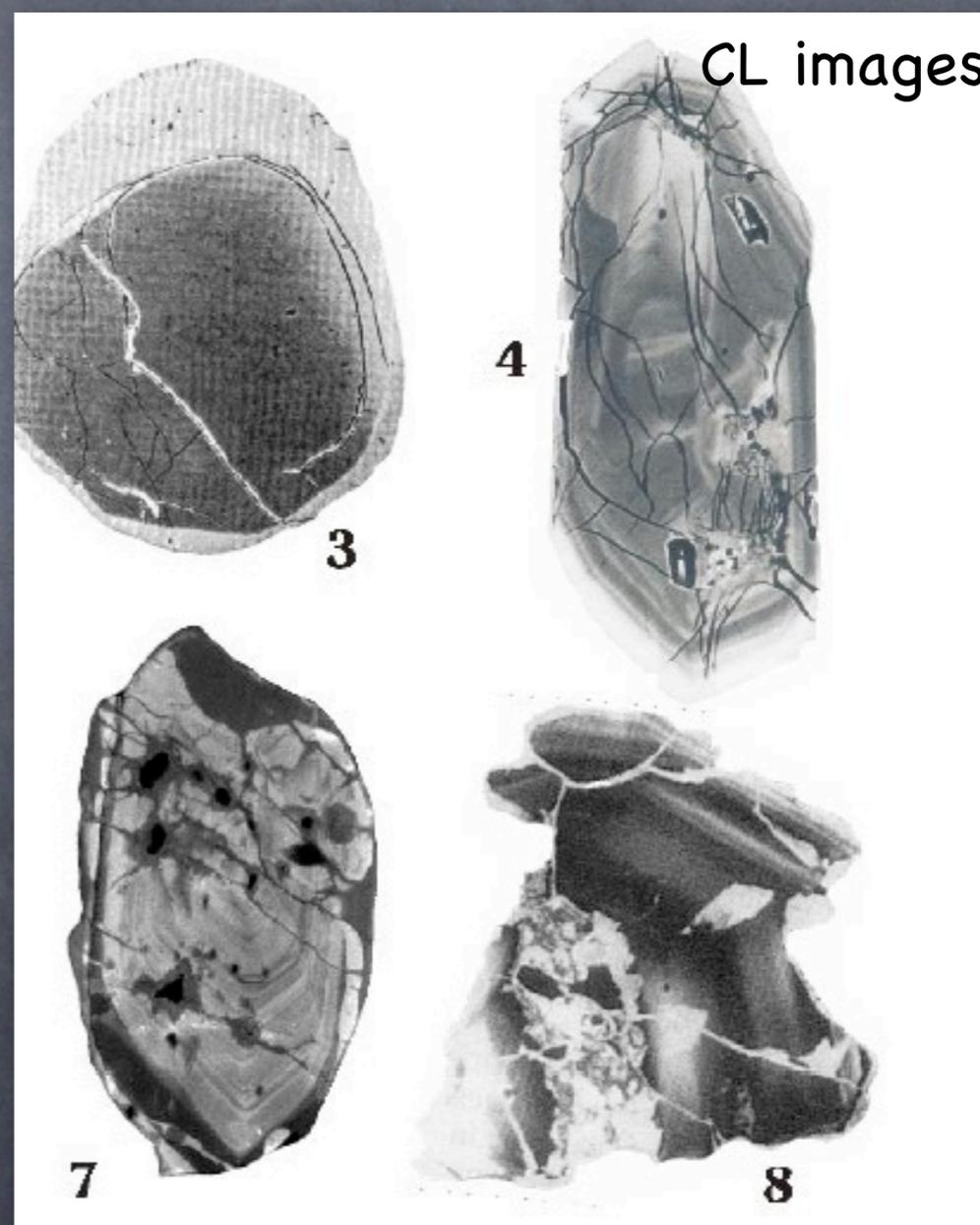
Xenocrystic cores in magmatic zircons

Zircon textures

High grade metamorphic rocks Highly fractured zircons

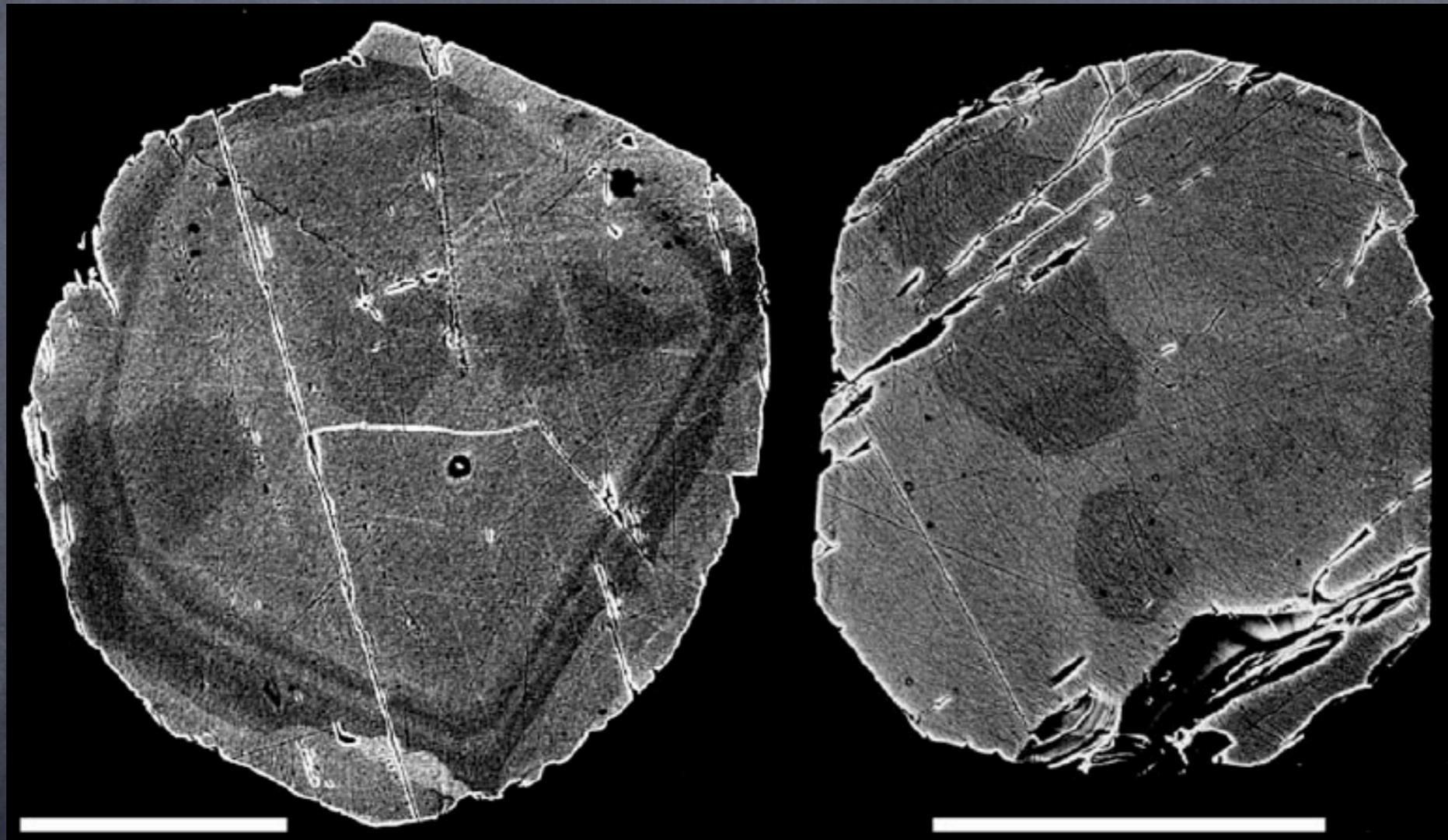


Corfu et al., RiMG 2003



Corfu et al., RiMG 2003

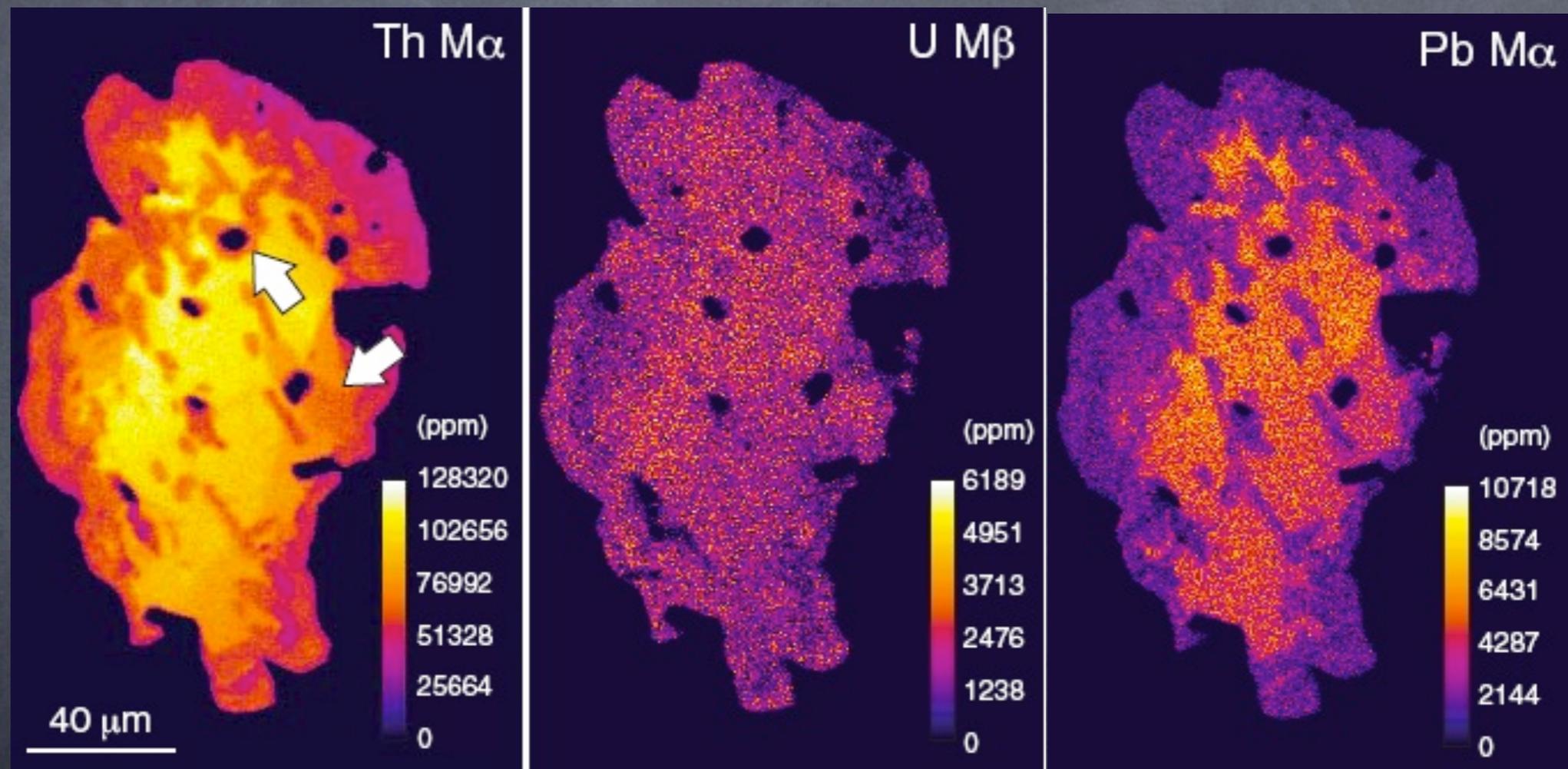
Monazite textures



Backscattered image

From Catlos et al., 2002

Monazite textures



From Goncalves et al., 2005

The analytical point of view

Required masses

^{204}Pb for common Pb correction

^{206}Pb ^{238}U $^{206}/^{238}$ age

^{207}Pb ^{235}U $^{207}/^{235}$ age

^{208}Pb ^{232}Th $^{208}/^{232}$ age

“Wish list”

- Submicron spatial resolution
 - nm scale depth profiling
 - Stoichiometric sampling
 - 100% transport efficiency
 - ppm precision and accuracy
- To distinguish age domains
- To avoid parent/daughter fractionation

In situ vs. bulk analysis

